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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/617,227	07/11/2003	Minoru Hasegawa	1082.1060	2346
21171	7590 04/06/2007		EXAMINER	
STAAS & HALSEY LLP SUITE 700 1201 NEW YORK AVENUE, N.W. WASHINGTON, DC 20005			HODGES, MATTHEW P	
			ART UNIT	PAPER NUMBER
			2879	
SHORTENED STATUTOR	Y PERIOD OF RESPONSE	MAIL DATE	DELIVERY MODE	
3 MO	NTHS	04/06/2007	PAPER	

Please find below and/or attached an Office communication concerning this application or proceeding.

If NO period for reply is specified above, the maximum statutory period will apply and will expire 6 MONTHS from the mailing date of this communication.

	Application No.	Applicant(s)	_			
	10/617,227	HASEGAWA ET AL.				
Office Action Summary	Examiner	Art Unit				
	Matt P. Hodges	2879				
The MAILING DATE of this communication ap Period for Reply	ppears on the cover sheet with the	correspondence address				
A SHORTENED STATUTORY PERIOD FOR REP WHICHEVER IS LONGER, FROM THE MAILING I - Extensions of time may be available under the provisions of 37 CFR 1 after SIX (6) MONTHS from the mailing date of this communication If NO period for reply is specified above, the maximum statutory period - Failure to reply within the set or extended period for reply will, by statu Any reply received by the Office later than three months after the maili earned patent term adjustment. See 37 CFR 1.704(b).	DATE OF THIS COMMUNICATION 136(a). In no event, however, may a reply be to divide apply and will expire SIX (6) MONTHS from the, cause the application to become ABANDON	N. imely filed In the mailing date of this communication. ED (35 U.S.C. § 133).				
Status						
1) Responsive to communication(s) filed on 16	February 2007					
	is action is non-final.					
· <u> </u>	•					
closed in accordance with the practice under	•					
Disposition of Claims						
4)⊠ Claim(s) <u>1-3,5-8,10-13 and 15-19</u> is/are pend	ling in the application					
4a) Of the above claim(s) is/are withdra						
5) Claim(s) is/are allowed.	awn nom consideration.					
6) Claim(s) <u>1-3,5-8,10-13 and 15-19</u> is/are reject	ted					
7) Claim(s) is/are objected to.	nou.					
8) Claim(s) are subject to restriction and/	or election requirement					
ordinary are subject to restriction and	or election requirement.					
Application Papers						
9)☐ The specification is objected to by the Examin	ner.					
10)⊠ The drawing(s) filed on 11 July 2003 is/are: a	ı)⊠ accepted or b)⊡ objected to	by the Examiner.				
Applicant may not request that any objection to the	e drawing(s) be held in abeyance. Se	ee 37 CFR 1.85(a).				
Replacement drawing sheet(s) including the corre	ction is required if the drawing(s) is of	bjected to. See 37 CFR 1.121(d).				
11) The oath or declaration is objected to by the E	Examiner. Note the attached Office	e Action or form PTO-152.				
Priority under 35 U.S.C. § 119						
12) Acknowledgment is made of a claim for foreig a) All b) Some * c) None of:		a)-(d) or (f).				
1. Certified copies of the priority documer						
2. Certified copies of the priority documer						
3. Copies of the certified copies of the pri		red in this National Stage				
application from the International Burea						
* See the attached detailed Office action for a lis	st of the certified copies not receiv	ed.				
Attachment(s)	_	•				
Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948)	4) Interview Summan Paper No(s)/Mail D					
2) ☐ Notice of Draftsperson's Patent Drawing Review (P1O-948) B) ☐ Information Disclosure Statement(s) (PTO/SB/08)	5) Notice of Informal					
Paper No(s)/Mail Date <u>1/23/2007</u> .	6) Other:					

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DETAILED ACTION

Response to Amendment

The Amendment, filed on 1/11/2007, has been entered and acknowledged by the Examiner.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-3, 5-7, 10-13, 15-17, and 19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Busio et al. (US 2001/0005115 A1) in view of Edwards et al. (US 6,037,280).

Regarding claims 1-3, 10, Busio discloses (see figure 1b) a color gas discharge panel including a rear substrate (7), barrier ribs (10), a phosphor layer (R, G, B), a front substrate (1) formed opposite the rear substrate and including in order, electrodes (3 and 4), an organic polymer first dielectric layer (5), an inorganic second dielectric layer (33), and a protection layer (6) of MgO. (Paragraph 21). Further the first dielectric layer is polysiloxane including a side chain of alkyl groups. (Paragraphs 0024-0030). The second dielectric layer between the organic dielectric layer and the protection layer is composed of a layer including ZrO₂ formed to block UV radiation from damaging the first dielectric layer and device. (Paragraph 0032). Busio does not appear to limit the material to ZrO₂ but also does not specify the use of an alternative material to perform the UV blocking function, however it has been held to be within the general

skill of a worker in the art to select a known material on the basis of its suitability for the intended use as a matter of obvious design choice. Edwards, in the field of UV blocking layers, discloses the known substitutes for ZrO₂ particulates in UV blocking materials including Al₂O₃. (See Column 4 lines 25-40). The substitution of one known UV blocking material for another known UV blocking material having similar properties is well within the skill of one of ordinary skill in the art. Further, the applicant fails to identify the use the stated materials absent ZrO₂ to solve any problem or yield any unexpected result that is not within in the scope of the teachings relied upon. Thus, it would have been obvious to one having ordinary skills in the art at the time the invention was made to have substituted Al₂O₃ as taught by Edwards in the place of ZrO₂ and into the device as disclosed by Busio since the selection of known materials for a known purpose is within the skill of the art.

Regarding claims 11-13 and 19, Busio discloses the device as claimed (see rejection of claims 1-3 and 10 above).

Further the limitation of that "the organic dielectric layer and the inorganic dielectric layer are formed together as a laminate" is drawn to a process of manufacturing which is incidental to the claimed apparatus. It is well established that a claimed apparatus cannot be distinguished over the prior art by a process limitation. Consequently, absent a showing of an unobvious difference between the claimed product and the prior art, the subject product-by-process claim limitation is not afforded patentable weight (see MPEP 2113).

Further regarding the limitation that the MgO layer is formed from an organic compound layer, it is well established that a claimed apparatus cannot be distinguished over the prior art by a process limitation. Consequently, absent a showing of an unobvious difference between the

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claimed product and the prior art, the subject product-by-process claim limitation is not afforded patentable weight. Specifically, the end product does not contain an organic compound (as it is burnt off by firing) thus the limitations drawn to the inclusion of an organic compound in the MgO layer have not been given patentable weight.

Regarding claims 5 and 15, the Al₂O₃ layer has a smaller bond distance between an oxygen atom and a Al atom than the wavelength of an atom vacuum ultra violet ray.

Regarding claims 6 and 16, the dielectric constant of Al₂O₃ (>9) is greater than that of organic dielectric layer.

Regarding claims 7 and 17, Busio further discloses the widths of the first and second dielectric layers being 10 μ m and 1 μ m respectively. (Paragraphs 0030 and 0032).

Claims 1-3, 5-8, 10-13 and 15-19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Aoki et al. (US 2003/0038599) in view of by Busio et al. (US 2001/0005115 A1) and further in view of Edwards et al. (US 6,037,280).

Regarding claims 1-3 and 10, Aoki discloses (see figure 1) a gas discharge panel including a rear substrate (21), barrier ribs (24), a phosphor layer (25), a front substrate (11) formed opposite the rear substrate, electrodes (12 and 13), a first dielectric layer (14), and a protection layer (15) of MgO: (Page 2 paragraph 34). Further the first dielectric layer is polysiloxane including a side chain of alkyl groups. (Page 2 paragraph 44) (Page 3 paragraphs 55 and 62). Aoki does not appear to disclose the use of a second dielectric layer between the organic first dielectric layer and the protection layer, however Busio, in the same field of endeavor, discloses the use of a thin inorganic dielectric layer made of ZrO₂ formed between the

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protection layer and the first dielectric layer. (Paragraph 0032). This second dielectric layer advantageously prevents degradation of the first dielectric layer by blocking UV radiation. Thus, it would have been obvious at the time the invention was made to a person having ordinary skills in the art to incorporate an inorganic second dielectric layer as taught by Busio into the gas discharge panel as disclosed by Aoki in order to advantageously prevent degradation of the first dielectric layer by blocking UV radiation.

Busio does not appear to limit the material to ZrO₂ but also does not specify the use of an alternative material to perform the UV blocking function, however it has been held to be within the general skill of a worker in the art to select a known material on the basis of its suitability for the intended use as a matter of obvious design choice. Edwards, in the field of UV blocking layers, discloses the known substitutes for ZrO₂ particulates in UV blocking materials including Al₂O₃. The substitution of one known UV blocking material for another known UV blocking material having similar properties is well within the skill of one of ordinary skill in the art. Further, the applicant fails to identify the use the stated materials absent ZrO₂ to solve any problem or yield any unexpected result that is not within in the scope of the teachings relied upon. Thus, it would have been obvious to one having ordinary skills in the art at the time the invention was made to have substituted Al₂O₃ as taught by Edwards in the place of ZrO₂ and into the device as disclosed by Busio since the selection of known materials for a known purpose is within the skill of the art.

Regarding claims 11-13 and 19, Aoki in view of Busio and further in view of Edwards discloses the device as claimed (see rejection of claims 1-3 and 10 above). Further the limitation of that "the organic dielectric layer and the inorganic dielectric layer are formed together as a

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laminate" is drawn to a process of manufacturing which is incidental to the claimed apparatus. It is well established that a claimed apparatus cannot be distinguished over the prior art by a process limitation. Consequently, absent a showing of an unobvious difference between the claimed product and the prior art, the subject product-by-process claim limitation is not afforded patentable weight (see MPEP 2113).

Further regarding the limitation that the MgO layer is formed from an organic compound layer, it is well established that a claimed apparatus cannot be distinguished over the prior art by a process limitation. Consequently, absent a showing of an unobvious difference between the claimed product and the prior art, the subject product-by-process claim limitation is not afforded patentable weight. Specifically, the end product does not contain an organic compound (as it is burnt off by firing) thus the limitations drawn to the inclusion of an organic compound in the MgO layer have not been given patentable weight.

Regarding claims 5 and 15, the second dielectric layer of Al_2O_3 has a bond distance that is smaller than the wavelength of an atom vacuum ultra violet array.

Regarding claims 6 and 16, the dielectric constant of polysiloxane is less than 3 while the dielectric constant of Al₂O₃ is greater than 9.

Regarding claims 7 and 17, the organic dielectric layer has a thickness of 15 μ m (Page 2 paragraph 41) while the inorganic layer is 1 μ m (Paragraph 0032).

Regarding claims 8 and 18, the protective layer is 1 μ m thick and is further formed from methods including sputtering and vapor deposition. The use of vapor deposition would lead to a porous body on the surface of the second dielectric layer. (Paragraphs 0071 and 0138)

Response to Arguments

Applicant's arguments filed 1/11/2007 have been fully considered but they are not persuasive.

Regarding applicant's assertion that the substitution of one material for another is not an obvious variation, the examiner respectfully disagrees. When the two materials in question are known in the art to perform the same function and are both used in some instances to perform that function, they will be considered art equivalent substitutions unless the applicant can provide some reason why they are not equivalent as cited. In this case, Busio distinguishes the UV blocking layer by its function and not specifically by its composition. The function of the layer is to protect the organic layer from damage from UV radiation emitted from the discharge cell. The prior art then anticipates the problem identified by the applicant. That the prior art only list one representative example of a radiation blocking material, will not be considered limiting when other radiation blocking materials are known and used as substitutes over the cited material.

Contact Information

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Matt P Hodges whose telephone number is (571) 272-2454. The examiner can normally be reached on 7:30 AM to 4:00 PM M-F.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Nimesh Patel can be reached on (571) 272-2457. The fax phone number for the organization where this application or proceeding is assigned is (703) 872-9306.

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